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Effect of Al Concentration on Microstructural Evolution of Fe-Cr-Al System: A Phase-field Approach

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Background

Fe-Cr-Al alloy

- Fe-Cr alloy is widely used in various fields.
- The addition of Al to Fe-Cr alloy has been considered as an effective method to improve the mechanical properties.

Strength

- Excellent corrosion resistance and high temperature strength.
- Excellent radiation tolerance and mechanical properties.



Kanthal APM FeCrAl alloy tube



SEM micrograph of the fracture surface of 21Cr-5AI (Jesper Ejenstam et al).

LWeakness

 The formation of α' phase, which generates embrittlement in the system.





- Phase-field model is thermodynamic and kinetics based model
- Quantitative modeling of phase transformations and microstructure evolutions.

LCalculation of phase diagram (CALPHAD) method

- CALPHAD method is a methodology to predict the system.
- It is used to simulate behavior of multicomponent phase.



Simulation Method

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CALPHAD Combined Phase-field modeling in ternary system

• Total free energy of Fe-Cr-Al system and Cahn-Hilliard equation. $F(r,t) = \int_{V} \left\{ \frac{1}{V_m} \left[G(c_{\text{Fe}}, c_{\text{Cr}}, c_{\text{Al}}) + \frac{1}{2} \kappa (\nabla c_{\text{Fe}})^2 + \frac{1}{2} \kappa (\nabla c_{\text{Cr}})^2 + \frac{1}{2} \kappa (\nabla c_{\text{Al}})^2 \right] \right\} dV$

 c_i : Concentration of i component.

Chemical free energy in ternary system.

 $G(c_{\text{Fe}}, c_{\text{Cr}}, c_{\text{Al}}) = c_{\text{Fe}}G_{Fe}^{0} + c_{\text{Cr}}G_{Cr}^{0} + c_{\text{Al}}G_{\text{Al}}^{0} + L_{FeCr}c_{\text{Fe}}c_{\text{Cr}} + L_{\text{CrAl}}c_{\text{Cr}}c_{\text{Al}} + L_{FeAl}c_{\text{Fe}}c_{\text{Al}} + RT[c_{\text{Fe}}ln(c_{\text{Fe}}) + c_{\text{Cr}}ln(c_{\text{Cr}}) + c_{\text{Al}}ln(c_{\text{Al}})]$

Onsager type mobility.

$$M_{Cr,Cr}(c_{Fe}, c_{Cr}, c_{Al}) = c_{Cr} [(1 - c_{Cr})^2 M_{Cr} + c_{Cr} c_{Al} M_{Al} + c_{Cr} c_{Fe} M_{Fe}]$$

$$M_{Al,Al}(c_{Fe}, c_{Cr}, c_{Al}) = c_{Al} [(1 - c_{Al})^2 M_{Al} + c_{Cr} c_{Al} M_{Cr} + c_{Al} c_{Fe} M_{Fe}]$$

$$M_{Cr,Al}(c_{Fe}, c_{Cr}, c_{Al}) = M_{Al,Cr} = c_{Cr} c_{Al} [c_{Fe} M_{Fe} - (1 - c_{Cr}) M_{Cr} - (1 - c_{Al}) M_{Al}]$$



Simulation Method

Semi-implicit Fourier spectral method Constraint

$$\frac{\partial c_{Cr}(r,t)}{\partial t} = \nabla \cdot \left[M_{Cr,Cr} \nabla \left(\frac{\delta F}{\delta c_{Cr}} \right) + M_{Cr,Al} \nabla \left(\frac{\delta F}{\delta c_{Al}} \right) \right]$$

$$\frac{\partial c_{\mathrm{Al}}(r,t)}{\partial t} = \nabla \cdot \left[\mathsf{M}_{\mathrm{Al},\mathrm{Al}} \nabla \left(\frac{\delta F}{\delta c_{\mathrm{Al}}} \right) + \mathsf{M}_{\mathrm{Al},\mathrm{Cr}} \nabla \left(\frac{\delta F}{\delta c_{\mathrm{B}}} \right) \right]$$

- 4th rank differential equation.
 \$\nabla^4\$
- The equation has severe time constraint. Δtκk⁴
- Concentration dependence of mobility. M_{i,j}(c_{Fe}, c_{cr}, c_{Al})

LSemi-implicit Fourier spectral scheme

$$\tilde{\mathbf{c}}_{A}^{n+1}(k,t) = \tilde{\mathbf{c}}_{A}^{n}(k,t) + \frac{\Delta tik}{(1+S_{A}\Delta t\kappa k^{4})} \cdot \{\mathbf{M}_{A,A} \times [ik'(\{\frac{\partial F}{\partial \mathbf{c}_{A}}\}_{k'}^{n} + 2\kappa k'^{2}\tilde{\mathbf{c}}_{A}^{n} + \kappa k'^{2}\tilde{\mathbf{c}}_{B}^{n})]_{r} + \mathbf{M}_{A,B} \times [ik'(\{\frac{\partial F}{\partial \mathbf{c}_{B}}\}_{k'}^{n} + 2\kappa k'^{2}\tilde{\mathbf{c}}_{B}^{n} + \kappa k'^{2}\tilde{\mathbf{c}}_{A}^{n})]_{r}\}_{k}$$

 We solved Cr and Al concentration over time, respectively.





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	Backgroun	Method	Result S	Future Work	Conclusio	Acknowled	lgem	en
Results								
2D simulation result								
As the Al concentration i	ncrease		(a)		(b)		1	Ę
 The difference of Cr an concentration between riched) phase and the phase increase. 	nd Al n the α ' (C α(Cr-depl	Cr- eted)					0.6 0.4 0.2 0 0.14	Cr concentratio
• The α ' phase fraction i	ncrease.	- 35Cr-9/ - 35Cr-7/ - 35Cr-5/ - 35Cr-3/ - 35Cr-0. - 35Cr-0. - 35Cr-0. - 35Cr-0. - 35Cr-0.	(C)	of the Cr or Al the average verage Al con	(d) I concentratio Cr concentrat icentration wa	n at t* = 5.0 > ion was 35% as (a)(c) 1% (l	 0.12 0.1 0.08 0.06 0.04 0.02 0 10⁶ and b)(c) 	Al concentration

Time Step(×10⁶)

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	Backgroun	Result	Future	Conclusio	Acknowledgemen
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Results

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2D simulation result

As the AI concentration increase

- The phase separation occur early compared to low average Al concentration.
- The peak of the α' phase number density increase.
- The number of the α' phase precipitates decrease significantly after the peak when the high average Al concentration..



	Backgroun d	Method	Result s	Future Work	Conclusio n	Acknowledgemen t
Results						

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3D simulation result

- Tendency for the α' phase forming an interconnected structure.
- In 2D simulation

 $P_i^{2D} = 1.0 - (1 - f_{\alpha'})^4.$

In 3D simulation

$$P_i^{3D} = 1.0 - \left(1 - f_{\alpha'}\right)^6.$$

 f_{α} : fraction of the α phase

 The tendency of the 3D simulation to form interconnected structure is higher than that of the 2D simulation.





	Backgroun d	Method	Result s	Future Work	Conclusio n	Acknowledgemen t
Results						

3D simulation result



Plot of the Cr concentration field at $t^* = 2.5 \times 10^6$. (a) 35Cr-1Al, (b)35Cr-1OAl, (c)40Cr-1Al, (d)40Cr-1OAl.



Plot of the Al concentration field at $t^* = 2.5 \times 10^6$. (e) 35Cr-1Al, (f)35Cr-1OAl, (g)40Cr-1Al, (h)40Cr-1OAl.



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Future Work

Incorporate nucleation model

- Classical nucleation theory.
- Poisson seeding algorithm





Effect of neutron irradiation on the microstructural evolution of the Fe-Cr-Al alloy

- Radiation-enhanced diffusion(RED) model.
- Cascade mixing model.





- The effect of Al concentration on the microstructural evolution of Fe-Cr-Al system was investigated by the phase-field model.
- We applied some methods that CALPHAD method, phase-field method, semi-implicit Fourier spectral method and GPU parallel computing scheme to simulate the microstructural evolution of Fe-Cr-Al alloy.
- An increase in the average Al concentration enhanced the initiation of phase separation, and the α' phase fraction.
- At 35Cr% and 40Cr%, the formation of an interconnected structure of the α' phase was observed in the 3D simulation, whereas the interconnected structure was not formed at the same chemical composition in 2D simulation.





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Thank you

