# Autoclave washout test to simulate exposure of the UO2 pellets under PWR conditions

Tae-Sik Jung <sup>a\*</sup>, Yeon-Su Na <sup>a</sup>, Min-Jae Ju <sup>a</sup>, Jung-Beom Kim<sup>a</sup>, Kwang-Young Lim <sup>a</sup>, Seung-Jae Lee <sup>a</sup> KEPCO Nuclear fuel, 1047 Daedeokdaero, Yuseong-gu Daejeon 30457, Republic of Korea <sup>\*</sup>Corresponding author: taesik@knfc.co.kr

## 1. Introduction

The development of UO<sub>2</sub> fuel containing additives capable of enhancing PCI resistance performance has been conducted worldwide for the past 30 years. For that purpose, it is aimed at weakening the adverse effect of the nuclear fuel itself on the safety of the cladding tube, not only in the high burnup but also in the load following operation. In particular, the creep test to evaluate the high temperature plasticity of UO<sub>2</sub> fuel was an important improvement which reduces stress to cladding tube [1,2]. However, not only the performance effect of maintaining the integrity of the cladding tube but also its oxidation behavior when exposed to the water chemistry environment in reactor has been evaluated [3]. The results of these characterization tests are important for the evaluation of its oxidation resistance, which are included in the topical reports for licensing of  $UO_2$  fuel which are doped with  $Cr_2O_3$ developed by Framatome and Al<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub> developed by GNF [4,5].

In the UO<sub>2</sub> pellets exposed to the oxidizing environment, the oxidation reaction occurs actively along the grain boundaries. At this time, the O/U ratio becomes higher and the crystal structure changes due to increasing oxygen ratio, resulting in fragmentation [6,7]. This phenomenon eventually accelerates from outside to inside of the pellets through grain boundaries. Therefore, it has been reported that the fragmentation of UO<sub>2</sub> sintered body can be prevented by reducing the grain boundary area vulnerable to oxidation or applying a second phase material having excellent oxidation resistance [3].

At present, KNF is developing PCI additives for enhancing PCI failure resistance of  $UO_2$ , and developed a washout autoclave system to evaluate oxidation resistance when  $UO_2$  containing additives is exposed to water chemistry environment in pressurized light water (PWR) reactor. In this work, we confirm that water chemistry and other conditions including reactor temperature and system pressure in primary system of PWR were simulated by using the autoclave washout equipment. In addition, the oxidation tendency of  $UO_2$ which varies with the sintering additive can be measured by weight increment.

## 2. Methods and Results

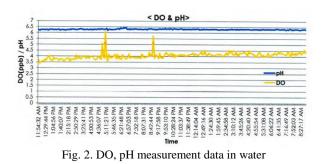
2.1 Washout autoclave system for simulating PWR water chemistry and its performance



Fig. 1. Schematic view of the washout equipment

The washout autoclave system was fabricated and installed as shown in Fig. 1 in order to evaluate the washout performance of  $UO_2$  exposed to the water chemistry environment after cladding failure. The loop system is a structure in which water is supplied from external water storage tank to the autoclave in which the corrosion test is performed. DO and pH sensors were installed to measure the water chemistry of the supplied water, and pressure sensor and thermocouple were installed in the autoclave to measure system temperature and pressure.

The washout autoclave equipment was run to verify whether it simulates the PWR water chemistry and environment conditions or not. As shown in Fig. 2. DO and pH of the water were measured to be maintained at less than 5 ppb and pH 6.4, which satisfied the PWR primary water chemistry standard. The temperature and pressure inside the autoclave were measured as shown in Fig. 4 at a set temperature and pressure of 18.5 MPa and 360°C respectively.



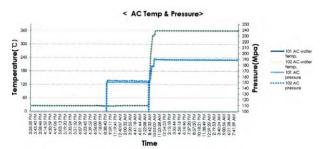


Fig. 3. Water temperature and pressure measurement data

2.2 UO<sub>2</sub> exposure in autoclave washout system

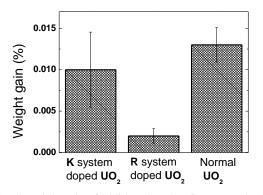


Fig. 4. Weight gain of additives doped  $\mathrm{UO}_2$  as a result of washout test

Figure 4 shows the results of an oxidation test using the autoclave washout equipment to evaluate the washout performance of UO<sub>2</sub> containing two (K, R) oxide additive systems under development. Five specimens were tested for each type and the results were measured by weight gain. The reason for analyzing the oxidation tendency through the weight increase measurement is that the O/U ratio changes to 2 (UO<sub>2</sub>), 2.33 (U<sub>3</sub>O<sub>7</sub>) and finally 2.66 (U<sub>3</sub>O<sub>8</sub>) as UO<sub>2</sub> is oxidized,

As a result, the weight increase of  $UO_2$  with R additive was 0.002% up to a factor of 5 ~ 6 times in comparison K system and general  $UO_2$ . However, typically  $UO_2$  oxidation occurs largely only when the dissolved oxygen level was more than 3 ppm [8]. The weight gain data in Fig. 4 are in agreement with the work of Une et al. [8]. Therefore, it is concluded that the rate of corrosion progression was slow during the experiment. Figure 5 shows the surface of the oxidized  $UO_2$  fuels. As can be estimated from the weight gain data in Figure 4, the surface oxidation of pellets appears not to be noticeable.

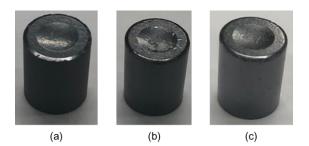


Fig. 5. External examination of samples oxidation test in PWR conditions, which are (a) K, (b) R system doped  $UO_2$  and (c) normal  $UO_2$ , respectively.

### **3.** Conclusions

Autoclave washout equipment of loop type was fabricated to measure the corrosion performance of  $UO_2$ in PWR water chemistry and environment conditions. The measured DO, pH, system pressure, and autoclave temperature satisfied the primary water chemistry. Oxidation resistance of  $UO_2$  containing K, R oxide additives has been evaluated using the equipment. As a result, the weight increase rate due to oxidation of  $UO_2$ with the R system were measured to have 5 to 6 times better corrosion resistance than the K added and general  $UO_2$ .

### REFERENCES

[1] X. Thibault et al., EDF PWR fuels – EDF operation experience, Water Reactor Fuel Performance Meeting, Proc.; TopFuel Paris, (2009), p.2153.

[2] C. Delafoy, AREVA NP  $Cr_2O_3$ -doped fuel development for BWRs, Light Water Reactor Fuel Performance Meeting, Proc. ANS Conf. San Francisco, (2007), p.1071

[3] C. Delafoy, IAEA-TECDOC 1654, Washout Behavior of Chromia-doped UO<sub>2</sub> and Gadolinia Fuels in LWR Environments, p 127.

[4] Licensing Topical Report ANP-10340NP, Incorporation of Chromia-Doped Fuel Properties in AREVA Approved Methods, (2016)

[5] Licensing Topical Report NEDO-33406, Additive Fuel Pellets for GNF Fuel Designs, (2009)

[6] B.J. Lewis, Fundamental aspects of defective nuclear fuel behavior and fission product release, Journal of Nuclear Materials 160, (1998), p.201-217

[7] C. Delafoy, P. Dewes, AREVA NP, Cr<sub>2</sub>O<sub>3</sub>-doped fuel development and qualification, Annual Meeting on Nuclear Technology, Proc. KTG Karlsruhe, (2007), p 504.

[8] K. Une, et al., Corrosion behavior of unirradiated oxide fuel pellets in high temperature water, Journal of Nuclear Materials 227, (1995), p. 32-39.