Validation of the Sodium Fire Models in the CONTAIN-LMR/1B-Mod.1 Code Against the ABCOVE Experiments

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Purpose and Ultimate Goal

Purposes : To validate the sodium fire models in the CONATIN-LMR/1B code against the ABCOVE experiments, and

To compare the results with the MELCOR results without using the sodium fire models. Ultimate Goal : Demonstration of the CONTAIN-LMR/1B code's capability to estimate transient P and T inside the containment during postulated accident conditions, as a design tool for the SFR(Sodium-cooled Fast Reactor) containment

Sodium Spray Fire Model

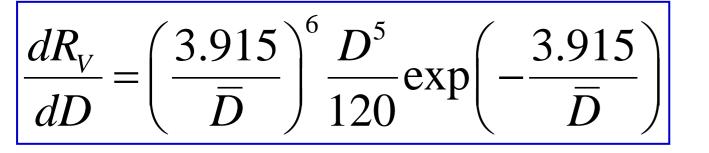
Sodium Pool Fire Model

- Phenomenological model used in NACOM code :
- Spray burning rate

$\dot{m}_{s}(t) = \int_{D_{i}} \int_{t'} \dot{m}_{f}(D(D_{i},t',t),V_{f}(D_{i},t',t)) d^{2}N$

Calculation begins by partitioning the injected sodium spray source among **11 discrete droplet-size**

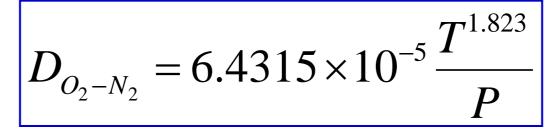
classes



- Now, d^2N can be derived from R_v = volume fraction of spray the sodium leak rate and drop size distribution
 - $d^{2}N = \frac{6m_{l}(t')}{\pi D_{i}^{3}\rho_{Na}} \frac{dR_{V}}{dD_{i}} dt' dD_{i}$
- Important user input parameters > Mass mean droplet diameter

- \dot{m}_{f} = sodium droplet burning rate = droplet diameter
- = droplet velocity d^2N = number of droplets in $D \sim D + dD \& z \sim z + dz$
- with droplets of smaller diameters than D
- = volume mean diameter

- Chemical reaction model used in SOFIRE II code : - Sodium burning rate is proportional to oxygen concentration at the sodium pool surface, to which oxygen in the atmosphere diffuse.
- Diffusion coefficient for oxygen-nitrogen mixtures



- **Sodium burning rate** $\left|\frac{dm}{dt} \cdot \frac{1}{A_{S}}\right| = H_{G}C\rho_{G}S$
- Gas transport coefficient, H_G , by heat-mass transfer analogy
 - $|H_G = 0.14 D_{O_2 N_2} | gS_C \frac{\beta}{n^2} (T_{SS} T_G)|$
 - **Default Assumptions in the CONTAIN-LMR/1B:**
- *P* = pressure [atm] *m* = mass of sodium burned = burning time $A_{\rm S}$ = sodium pool surface area **C** = mass fraction of oxygen $\rho_{\rm G}$ = gas density **S** = stoichiometric combustion ratio = gravity

T = gas (film) temperature [K]

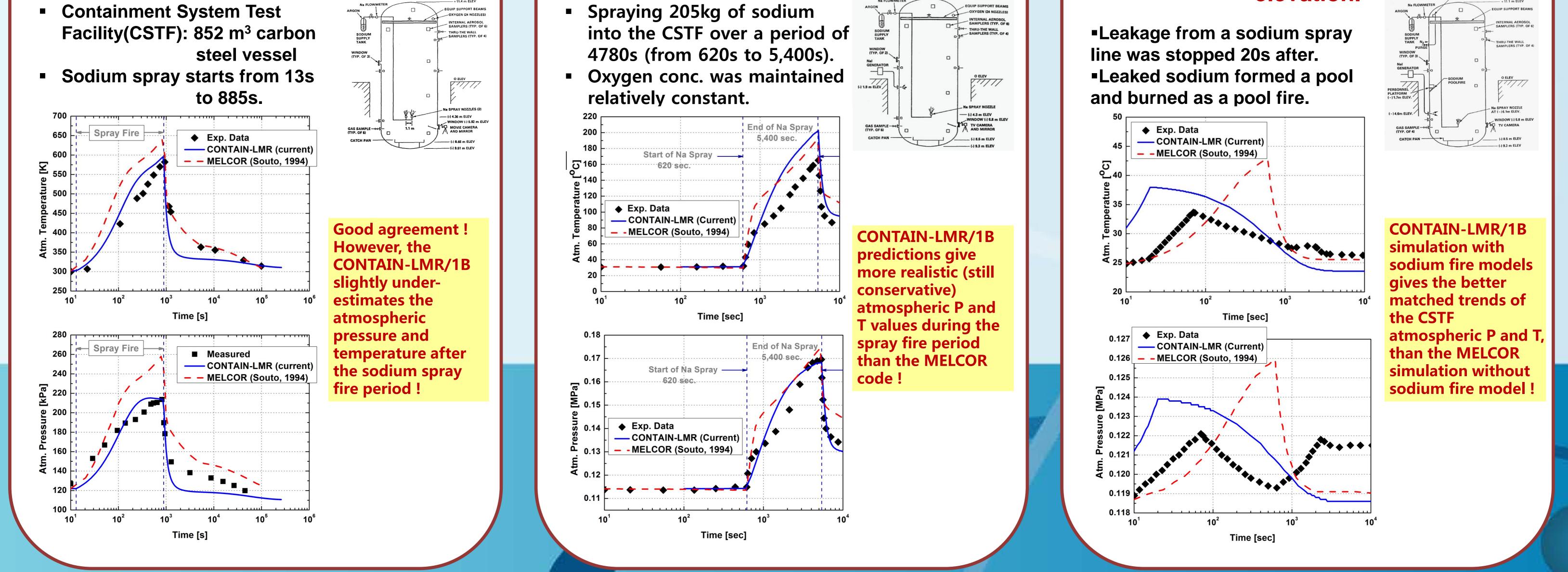
- = gas expansion coefficient
- = kinematic viscosity,
- $T_{\rm SS}$ = sodium surface temp.
- $T_{\rm G}$ = gas temperature
- Sc = Schmidt number

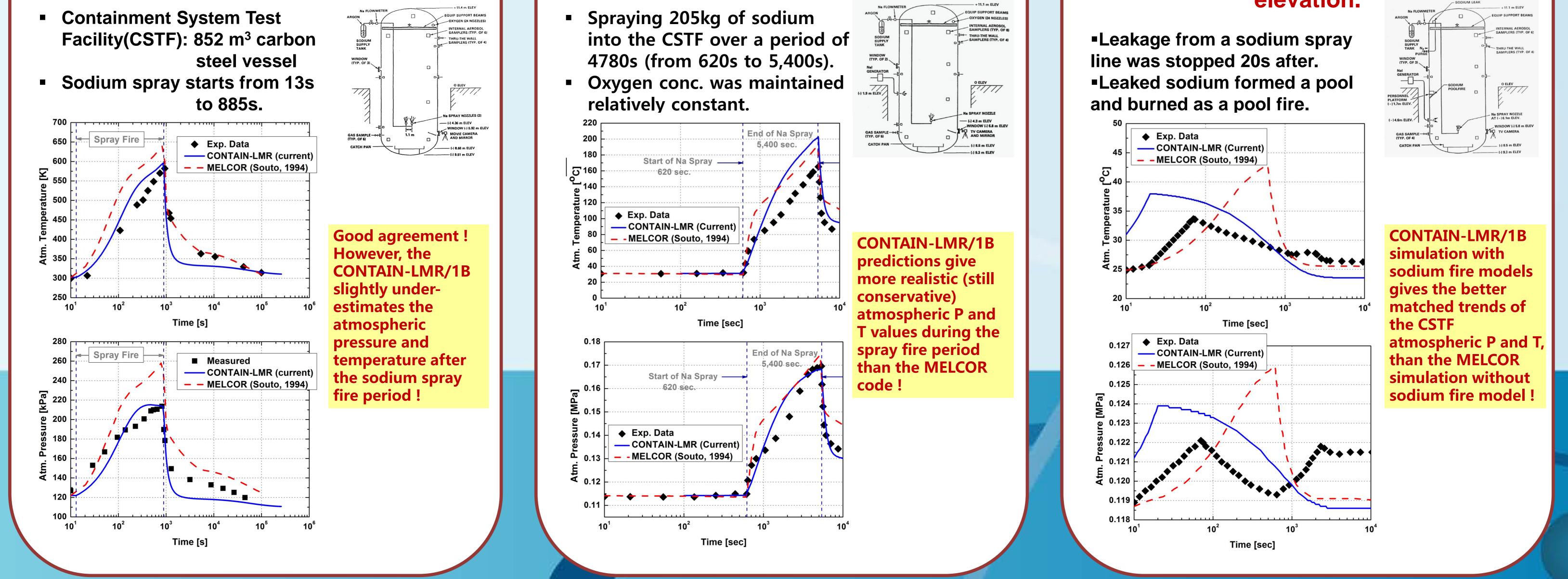
- > Fall height and terminal velocity
- \succ Relative proportions of Na₂O and Na₂O₂ in the reaction products

- > All the produced peroxide(Na_2O_2) is aerosolized.
- > All the produced monoxide(Na₂O) enter the pool.

Validation against ABCOVE AB5

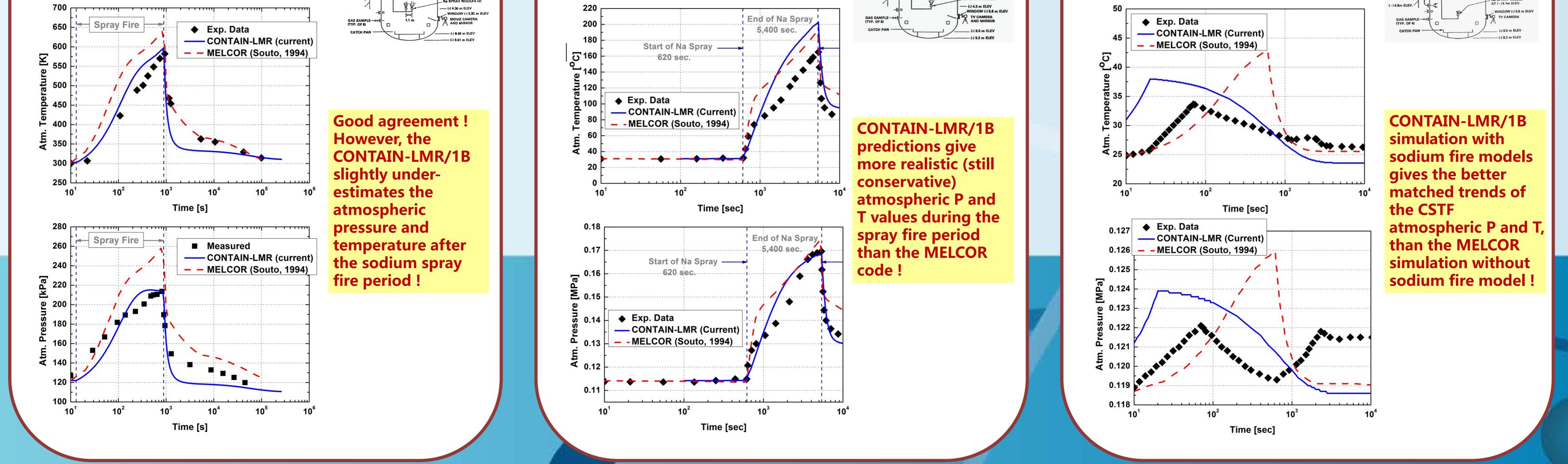
Single-species aerosol test by spraying sodium into air





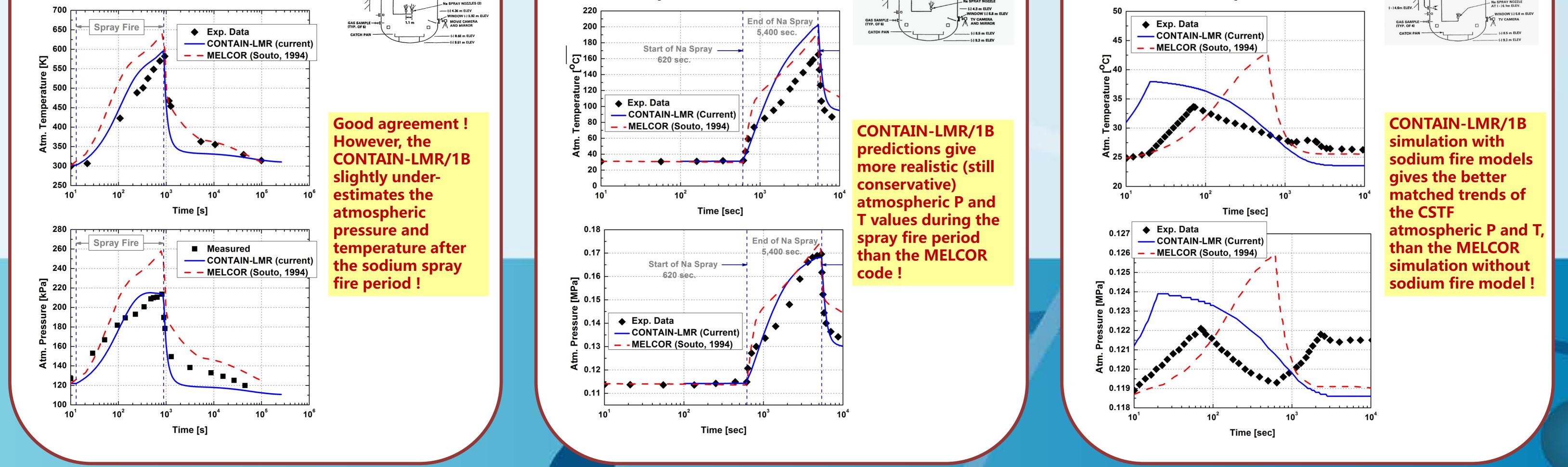
Validation against ABCOVE AB6

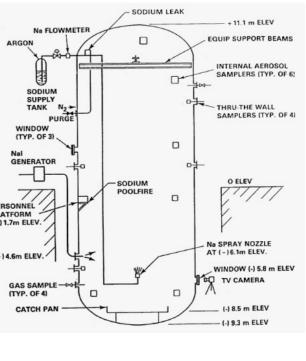
Nal aerosol release test in the presence of a sodium spray fire



Validation against ABCOVE AB7

A small sodium pool fire following sodium leakage from a line at 10-m elevation.





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